

1 **Recombinant Sox enzymes from *Paracoccus pantotrophus* degrade hydrogen**
2 **sulfide, a major component of oral malodor**

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13 **Topic of the manuscript:**

14 For the first time we report our findings on the H₂S-degrading activity of a mixture of
15 seven recombinant Sox enzymes from *Paracoccus pantotrophus* with respect to oral
16 malodor prevention. Oral malodor is a common problem, but effective methods to

17 control it have not been established. Major malodorous components are volatile sulfur
18 compounds including H₂S. We show the possibility of applying bacterial Sox system
19 to oral malodor prevention.

20

21 **Running headline:** *P. pantotrophus* rSox enzymes degrade H₂S

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25

26 **Abstract**

27 Hydrogen sulfide (H₂S) is emitted from industrial activities, and several chemotrophs
28 possessing Sox enzymes are used for the removal of H₂S. In dental field, oral malodor
29 is a common problem and major malodorous components are volatile sulfur compounds
30 (VSCs), including H₂S and methyl mercaptan. *Paracoccus pantotrophus* is an aerobic,
31 neutrophilic facultatively autotrophic bacterium that has sulfur-oxidizing (Sox) enzymes
32 to use sulfur compounds as an energy source. In this study, we cloned Sox enzymes of
33 *P. pantotrophus* GB17 and evaluated their VSC-degrading activity with respect to oral
34 malodor prevention. Six genes, *soxX*, *soxY*, *soxZ*, *soxA*, *soxB*, and *soxCD*, were
35 amplified from *P. pantotrophus* GB17. Each fragment was cloned into a vector for the
36 expression of 6×His-tagged fusion proteins in *Escherichia coli*. Recombinant Sox
37 (rSox) proteins were purified from whole-cell extracts of *E. coli* using nickel affinity
38 chromatography. The enzyme mixture was investigated for the degradation of VSCs
39 using gas chromatography. Each of the rSox enzymes was purified to apparent
40 homogeneity, as confirmed by SDS-PAGE. The rSox enzyme mixture degraded H₂S
41 in dose- and time-dependent manners. All rSox enzymes were necessary to degrade

42 H₂S. The H₂S-degrading activity of rSox enzymes was stable at 25-80°C, and the
43 optimum pH was 7.0. The amount of H₂S produced by periodontopathic bacteria or
44 oral bacteria collected from human subjects decreased after incubation with rSox
45 enzymes. These results suggest that the combination of rSox enzymes from *P.*
46 *pantotrophus* GB17 could be useful for the prevention of oral malodor.

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48 **Key words:** hydrogen sulfide, oral bacteria, oral malodor, *Paracoccus pantotrophus*,
49 sulfur-oxidizing enzyme

50

51 Introduction

52 Hydrogen sulfide (H_2S) is emitted from industrial activities, especially from biogas and
53 protein-rich industrial wastes. H_2S has high toxicity and is very corrosive to internal
54 combustion engines. Currently, many commercial chemical technologies are used for
55 H_2S removal in industry (41). However, these chemical H_2S removal processes are
56 expensive due to high chemical requirements, energy, and disposal costs, and they tend
57 to have short-term effects. Biological treatment methods for H_2S removal are
58 desirable, and several studies have investigated this (10, 32). The oxidation of H_2S is
59 mediated by various aerobic lithotrophic and anaerobic phototrophic bacteria (35).

60 In dental fields, oral malodor, the presence of unpleasant or foul-smelling
61 breath, is a common problem throughout the world and can cause significant social or
62 psychological effects on those who are suffering from it (6, 15, 21). Oral malodor is a
63 mixture of malodorous components, including volatile sulfur compounds (VSCs), such
64 as H_2S , methyl mercaptan (CH_3SH), and dimethyl sulfide (6, 15). VSCs are produced
65 in saliva, gingival crevices, on the tongue surface, and in other areas via the putrefactive
66 activities of microorganisms on sulfur-containing amino acids, such as cysteine and

67 methionine (3, 6). H₂S and CH₃SH are primarily responsible for oral malodor and
68 comprise approximately 90% of the VSC content in mouth air (43).

69 VSCs also have adverse effects on oral tissues. It has been shown that H₂S
70 damages gingival epithelial cells (42), increases the permeability of oral mucosa *in vitro*
71 (24), and causes apoptosis in human gingival fibroblasts (2). H₂S was predominantly
72 detected in the pockets (27) associated with periodontal bacteria, including
73 *Fusobacterium nucleatum*, one of the most active oral bacteria to produce H₂S from
74 L-cysteine (26). Many reports have also demonstrated that the presence of CH₃SH is
75 involved in the induction or progression of periodontal disease. Exposure to CH₃SH
76 inhibits cell migration in periodontal ligament cells (18), epithelial cell growth and
77 proliferation (38). *Porphyromonas gingivalis*, a black-pigmented anaerobic bacterium
78 and a major pathogen in adult periodontitis, produces large amounts of CH₃SH in
79 human serum (26) from L-methionine.

80 There have been several attempts to prevent and reduce oral malodor (6, 40, 45).
81 Mechanical prophylaxis using a toothbrush and a tongue scraper is a basic method to
82 remove bacterial cells and substrates from the oral cavity, but the organisms still grow

83 and accumulate. Mouth rinses containing antimicrobial agents, such as chlorhexidine
84 and cetylpyridinium chloride, have been used to reduce bacterial numbers, leading to a
85 reduction in oral malodor (9, 47). Additionally, zinc ions have been used to reduce
86 VSCs through their oxidizing effects on the thiol groups in VSC precursors (1, 16).
87 However, the effects of specifically formulated mouth rinses for treating oral malodor
88 are, in general, unclear (40). Therefore, new methods to control VSCs in the oral
89 cavity should be established for the prevention of oral malodor.

90 The purpose of this study was to examine the possibility of applying
91 sulfur-oxidizing (Sox) enzyme system to control oral malodor. We chose the
92 sulfur-oxidizing bacterium, *Paracoccus pantotrophus*, which is an aerobic,
93 Gram-negative, neutrophilic facultatively autotrophic bacterium that grows using
94 thiosulfate or molecular hydrogen as an energy source and heterotrophically using a
95 large variety of carbon sources (13). The Sox enzyme system in *P. pantotrophus* has
96 been studied so far. The *sox* gene region of *P. pantotrophus* comprises 12 open
97 reading frames and seven genes, *soxXYZABCD*, encode proteins essential for sulfur
98 oxidation in vitro (34). The Sox proteins of *P. pantotrophus* are located in the

99 periplasm (12), and four proteins, SoxXA, SoxYZ, SoxB, and SoxCD, are required for
100 H₂S-, sulfur-, thiosulfate-, and sulfite-dependent horse cytochrome *c* reduction (34).

101 We report the cloning of *sox* genes from *P. pantotrophus* GB17 – *soxX*, *soxY*, *soxZ*,
102 *soxA*, *soxB*, and *soxCD* – and the characteristics of recombinant Sox (rSox) enzymes.
103 We investigated rSox enzymatic degradation of VSCs produced by periodontopathic
104 bacteria and oral bacteria.

105

106 **Materials and methods**

107 *Bacterial strains and culture conditions*

108 All strains were cultivated at 37°C. *P. pantotrophus* GB17 (NBRC 102493) was
109 obtained from NBRC (Kisarazu, Japan) and used throughout this study. Seed cultures
110 were grown aerobically in brain heart infusion (BHI; Becton, Dickinson and Company,
111 Sparks, MD, USA) broth supplemented with 4 mM magnesium sulfate. *Escherichia*
112 *coli* XL II was grown aerobically in Luria Bertani (LB; Difco Laboratories, Detroit, MI,
113 USA) medium. *P. gingivalis* W83 was grown anaerobically in GAM broth (Nissui
114 Medical Co., Tokyo, Japan) supplemented with hemin (5 µg mL⁻¹) and menadione (1 µg

115 mL⁻¹). *F. nucleatum* ATCC10953 was grown anaerobically in BHI broth
116 supplemented with 5 mg mL⁻¹ yeast extract and 0.3 mg mL⁻¹ cysteine-HCl. Ampicillin
117 (100 µg mL⁻¹) was added when appropriate.

118

119 *DNA manipulation*

120 Standard DNA recombinant procedures, such as DNA isolation, restriction
121 endonuclease digestion, ligation, transformation of competent *E. coli* cells, and agarose
122 gel electrophoresis, were carried out as described by Sambrook *et al.* (37).

123 Chromosomal DNA was isolated from *P. pantotrophus* GB17 cells using the Dr.

124 GenTLE (from yeast) High Recovery DNA extraction kit (Takara Bio Inc., Shiga,

125 Japan).

126

127 *DNA amplification*

128 To improve the fidelity of the PCR assay for *soxX*, *soxY*, *soxZ*, *soxA*, *soxB*, and *soxCD*

129 genes, we used Tks Gflex DNA polymerase (Takara Bio Inc.). The reaction mixture

130 (50 µL total) contained 25 µL of 2× Gflex PCR buffer (containing 2 mM of Mg²⁺ and

131 400 μ M dNTPs; Takara Bio Inc.), 1 μ L of Tks Gflex DNA polymerase (Takara Bio
132 Inc.), 0.01 nM each primer, 1 μ L DNA template, and the volume was adjusted with
133 nuclease-free water (Roche Diagnostics, Indianapolis, IN). The reaction was carried
134 out for 30 cycles under the following conditions: initial denaturation at 94°C for 1 min,
135 denaturation at 98°C for 10 s, annealing at 48°C for 15 s, and extension at 68°C for 2.5
136 min.

137
138 *Cloning of sox genes from P. pantotrophus GB17*

139 Six *sox* genes (*soxX*, *soxY*, *soxZ*, *soxA*, *soxB*, and *soxCD*) were amplified by PCR from
140 the 5'-terminus of the *sox* gene (13 kbp): *soxX* fragment (474 bp), *soxY* fragment (423
141 bp), *soxZ* fragment (330 bp), *soxA* fragment (873 bp), *soxB* fragment (1695 bp), and
142 *soxCD* fragment (2431 bp). These short double-stranded DNAs were compatible with
143 a *Bam*HI site on the 5'-end and a *Hind*III site on the 3'-end. The primers were
144 designed to create *Bam*HI and *Hind*III restriction sites (underlined) within the PCR
145 product (Table 1). The pQE30 vector (Qiagen, Tokyo, Japan) was used for
146 construction of histidine-tagged recombinant proteins. Each DNA fragment,

147 containing a *sox* gene, was digested with *Bam*HI and *Hind*III and ligated into the pQE30
148 vector for the expression of 6×His-tagged fusion proteins. To obtain the Sox products
149 of *P. pantotrophus* GB17, *E. coli* XL II competent cells were transformed with each
150 resulting plasmid (*soxX*, *soxY*, *soxZ*, *soxA*, *soxB*, and *soxCD*). Positive colonies were
151 selected and re-plated on tryptic soy (TS; Becton, Dickinson and Company) agar
152 containing ampicillin (100 µg mL⁻¹). The nucleotide sequences of the inserted
153 fragments were confirmed by PCR to verify that the fragments were correct and did not
154 contain nucleotide substitutions or deletions. BLAST nucleotide sequence analysis
155 was also performed for DNA sequence identification.

156

157 *Preparation of recombinant Sox proteins*

158 The transformants were grown in LB medium with ampicillin (100 µg mL⁻¹) at 37°C
159 until the optical density at 550 nm (OD₅₅₀) reached 0.5.
160 Isopropyl-β-D(-)-thiogalactopyranoside was added to the culture at a final concentration
161 of 1 mM, and the cultures were grown for an additional 4 h. The cells were harvested
162 by centrifugation (5,000 × *g*, 15 min, 4°C) and lysed using an ultrasonic sonifier

163 equipped with a microtip (Model W-220-F; Heat Systems Ultrasonics Inc., Plainview,
164 NY, USA) on ice. The cell extract was obtained by centrifugation ($10,000 \times g$, 15
165 min, 4°C) and subjected to Ni-NTA resin (Qiagen) affinity column chromatography.
166 The purification procedures followed the manufacturer's instructions, and the purity of
167 the recombinant Sox proteins was analyzed by SDS-PAGE. Eluted proteins were
168 refolded by sequential dialysis against a urea-decreasing phosphate buffer at 4°C (48).
169 The amounts of proteins were determined using the Lowry method with bovine serum
170 albumin as the standard (19).

171

172 *SDS-PAGE and Western blotting*

173 SDS-PAGE was performed using 15% polyacrylamide gels according to the method of
174 Laemmli (17). After electrophoresis, the gel was stained with Coomassie brilliant blue
175 R-250. The low molecular weight electrophoresis calibration kit (Amersham
176 Pharmacia Biotech, Uppsala, Sweden) was used for molecular mass markers. For
177 Western blotting, the proteins subjected to SDS-PAGE were transferred
178 electrophoretically to a nitrocellulose membrane according to the method of Burnette

179 (8). After blocking with 2% skimmed milk in Tris-buffered saline (20 mM Tris, 150
180 mM NaCl, pH 7.2) containing 0.1% Triton X-100 (TBS-T), the membrane was reacted
181 with HRP-conjugated mouse anti-6×His monoclonal antibody (Wako Pure Chemical
182 Industries Ltd., Osaka, Japan). The membrane was washed three times with TBS-T,
183 and fluorescence detection was performed using a chemiluminescence detection system
184 (Clarity western ECL blotting, Bio-Rad Laboratories, Inc., Hercules, CA).

185

186 *Enzyme assay*

187 We investigated the ability of a rSox enzymes mixture reconstituted from rSoxX,
188 rSoxY, rSoxZ, rSoxA, rSoxB, and rSoxCD to degrade H₂S generated from NaHS. The
189 assay mixture (1 mL) for the determination of H₂S-degrading activity contained various
190 amounts of *P. pantotrophus* GB17 cells or rSox enzymes and 20 nmol of sodium
191 hydrogen sulfide (NaHS) to generate H₂S in 10 mM phosphate buffer (pH 7.0). All
192 the reactions were performed in sterile 15-mL polypropylene tubes sealed with a
193 silicone plug. The NaHS was added to start the reaction. After incubation for 2 h at
194 37°C, a sample (2.5 mL) of the vapor above the assay mixture in the tube was removed

195 using a gas-tight syringe and analyzed by gas chromatography (model GC-2014;
196 Shimadzu Corp., Kyoto, Japan) using a glass column packed with 25% β ,
197 β^2 -oxydipropionitrile on a 60-80-mesh Chromosorb W AW-DMCS-ST device
198 (Shimadzu Corp.) fitted with a flame photometric detector at 70°C. The concentration
199 of VSCs was determined using standard H₂S and CH₃SH gas prepared with a Permeater
200 PD-1B (Gastec, Ayase, Japan).

201 The assay mixture (1 mL) used to determine the rSox enzyme activity to degrade
202 VSCs produced by periodontopathic bacteria were prepared following the method of
203 Yoshimura *et al.* (49). Briefly, bacterial strains were grown at 37°C until an OD₅₅₀ of
204 about 0.6 was reached. The cells were harvested and washed three times with a
205 buffered salt solution (40 mM potassium phosphate buffer, 50 mM NaCl, pH 7.7).
206 The cells were suspended in the buffered salt solution to an OD₅₅₀ of 0.3 for *F.*
207 *nucleatum* ATCC10953 and OD₅₅₀ of 1.0 for *P. gingivalis* W83. To determine H₂S, a
208 reaction mixture was prepared consisting of 100 μ L of *F. nucleatum* ATCC10953 or *P.*
209 *gingivalis* W83 cell suspension and 0.5 nmol each of rSoxX, rSoxY, rSoxZ, rSoxA,
210 rSoxB, and rSoxCD, and buffered salt solution was added to a tube and sealed with a

211 silicone plug. The reaction was initiated by adding 30 μ L of 33 mM L-cysteine. For
212 the determination of CH_3SH , 30 μ L of 33 mM L-methionine were added instead of
213 L-cysteine. The assay mixtures were kept at 37°C and after a 2-h incubation, the
214 reactions were stopped by adding 500 μ L of 3 M phosphoric acid. Then, 10 min later,
215 a sample (2.5 mL) of the vapor above the assay mixture was analyzed by gas
216 chromatography, as described above.

217 The assay mixture used to determine the degrading activity of H_2S produced by
218 oral bacteria was prepared as described by Tonzetich and Johnson (44) with
219 modifications. Paraffin-stimulated whole saliva was collected from three healthy and
220 non-smoking participants (28-58 years of age), which was approved by the ethics
221 committee of Kagoshima University (authorization number 572). The bacterial
222 sediment was collected by centrifugation ($10,000 \times g$, 15 min, 4°C) and washed three
223 times with buffered salt solution. The bacterial sediment was disrupted by
224 ultrasonication on ice and suspended in the buffered salt solution to an OD_{550} of 1.0 to
225 measure the extracellular H_2S produced from L-cysteine by oral bacteria. The reaction
226 mixture (3 mL) consisted of 2 mL bacterial suspension and 0.5 nmol each of rSoxX,

227 rSoxY, rSoxZ, rSoxA, rSoxB, and rSoxCD in buffered salt solution. The subsequent
228 procedures were as described above.

229 To evaluate heat stability, rSox enzymes were mixed and heated at 25 to 100°C
230 for 30 min and subjected to the H₂S-degrading assay. To examine the effect of pH on
231 rSox enzyme activity, 10 mM phosphate buffer with pH 4 to 9 was prepared.

232

233 *Statistical analysis*

234 Data were averaged for three independent experiments. Statistical analysis was
235 performed using Student's *t*-test or one-way analysis of variance (ANOVA) followed by
236 Dunnett's test. *P*-values < 0.05 were considered to indicate statistical significance.

237

238 **Results**

239 *Purification of recombinant Sox proteins*

240 Figure 1A shows purified recombinant proteins. rSoxX migrated at 23 kDa; rSoxY at
241 18 kDa; rSoxZ at 21, 22, and 43 kDa; rSoxA at 43 kDa; and rSoxB at 68 kDa (Fig. 1A).

242 In Western blot analysis, the anti-6×His antibody reacted with the rSoxX, rSoxY,

243 rSoxZ, rSoxA, and rSoxB proteins (Fig. 1B). rSoxCD was separated into two bands
244 (51 and 54 kDa) by SDS-PAGE, and the 54-kDa band reacted with the anti-6×His
245 antibody. Because the 6×His was attached to rSoxC, we suggest that the 51-kDa band
246 was rSoxD and the 54-kDa band rSoxC. rSoxZ showed three bands and the antibody
247 to 6×His reacted with all of these components, so we suggest that the 43-kDa band was
248 a dimer, and the 21-kDa band lacked some portion of rSoxZ (the 22-kDa band).

249

250 *P. pantotrophus GB17 activity to degrade H₂S*

251 Whole cells of *P. pantotrophus* GB17 were examined for degradation of H₂S generated
252 from NaHS. *P. pantotrophus* GB17 cells showed H₂S-degrading activity, and the H₂S
253 concentration decreased in a cell number-dependent manner (Fig. 2). Approximately
254 50% of H₂S was degraded when a bacterial suspension containing 1.0×10^7 cells was
255 used.

256

257 *rSox enzyme activity to degrade H₂S*

258 A combination of the six rSox proteins (rSoxX, rSoxY, rSoxZ, rSoxA, rSoxB, and
259 rSoxCD) was investigated for its ability to degrade H₂S generated from NaHS. After
260 incubation, the rSox enzymes mixture showed degradation of H₂S in a dose-dependent
261 manner (Fig. 3A). The concentration of H₂S decreased significantly and reached a
262 stable level after 0.125 nmol of proteins were used. In time-course experiments, the
263 ratio of H₂S concentration (sample/control) decreased to approximately 45% at 2-h
264 incubation and was stable thereafter (Fig. 3B). In the thermal stability study, the
265 amount of H₂S in the assay mixture increased gradually as the temperature was raised
266 (Fig. 4A). However, the amount of H₂S was still significantly lower than that in the
267 control even after heating at 80°C. After heating at 100°C, there was no difference in
268 the amount of H₂S compared with the control. Regarding the effect of pH on rSox
269 enzyme activity, the percentage of H₂S concentration (sample/control) was lowest at pH
270 7.0, and higher values were observed at other pH values (Fig. 4B).

271

272 *Effect of each component of the rSox enzyme mixture on H₂S degradation*

273 Table 2 shows the effects of the components of the rSox enzyme mixtures on
274 degradation of H₂S. In this experiment, the rSoxX and rSoxA or rSoxY and rSoxZ
275 pairs were omitted from the assay, because those pairs of enzymes work together in the
276 Sox system (12, 34). When a mixture of rSox enzymes reconstituted from all rSox
277 enzymes was used, the H₂S level decreased significantly compared with the control (no
278 rSox enzymes). In the other cases, omitting certain rSox enzymes from the assay, H₂S
279 levels were not significantly different from the control, and the H₂S ratios relative to
280 control levels were approximately 70%. Bovine serum albumin, which was used as a
281 non-sulfur-oxidizing control, showed no reducing effect of H₂S.
282
283 *rSox enzyme activity to degrade VSCs produced by periodontopathic bacteria and oral*
284 *bacteria*
285 We first determined the ability of rSox enzymes to degrade H₂S produced by *F.*
286 *nucleatum* ATCC10953 and *P. gingivalis* W83. The rSox enzyme mixture
287 significantly degraded the H₂S produced by the two bacterial strains (Table 3). We
288 also determined CH₃SH degradation by the rSox enzyme mixture and found that it

289 could decrease the CH₃SH level, but that the difference was not significant compared
290 with the control (no enzymes). Next, oral bacteria were collected from human saliva
291 and used to produce H₂S. The rSox enzyme mixture significantly degraded H₂S
292 produced from the oral bacteria from all subjects, although the degree differed among
293 the subjects (Table 4).

294

295 **Discussion**

296 In this study, we firstly demonstrated the degradation of H₂S by the rSox enzymes from
297 *P. pantotrophus* GB17 in view of oral malodor prevention. We produced rSox
298 enzymes to examine their VSC-degrading activity. We first attempted to construct
299 rSoxXA, rSoxYZ, rSoxB, and rSoxCD proteins, but only rSoxB and rSoxCD proteins
300 were obtained successfully. Thus, we next constructed each enzyme as rSoxX, rSoxY,
301 rSoxZ, and rSoxA. We demonstrated that the combination of rSoxX, rSoxY, rSoxZ,
302 rSoxA, rSoxB, and rSoxCD proteins degraded H₂S, and the activity was almost constant
303 after reaching the maximum effect (Fig. 3A). Regarding the use of recombinant Sox
304 enzymes, rSoxXA expressed in *E. coli* was found to be functional in the reconstituted

305 enzyme system to activate H₂S-dependent cytochrome *c* reduction (35). This report is
306 the first to describe a Sox system composed of each recombinant Sox protein for
307 degrading H₂S.

308 High cytochrome *c*-reducing activity by the Sox system was observed using H₂S
309 as the substrate (34), and to achieve optimal activity, SoxXA, SoxYZ, SoxB, and
310 SoxCD were required (12, 34). In the present study, we demonstrated that all of the
311 enzymes – rSoxX, rSoxY, rSoxZ, rSoxA, rSoxB, and rSoxCD – were necessary for H₂S
312 degradation (Table 2). The central protein of the Sox enzyme system is SoxYZ, to
313 which the sulfur substrate is covalently linked, oxidized, and finally released as sulfate
314 by three other Sox proteins (11). SoxXA binds the sulfur substrate and covalently
315 links it to the thiol of the single cysteine 110 residue of the SoxY subunit. SoxY forms
316 a complex with SoxZ. The outer sulfur atom is oxidized by sulfur dehydrogenase
317 SoxCD, yielding SoxY-cysteine-S-sulfate. Finally, SoxY-cysteine-S-sulfate is
318 hydrolyzed by the dimanganese SoxB protein to yield sulfate and to regenerate SoxY
319 for a new reaction cycle of SoxYZ (13, 36). When SoxB or SoxYZ was omitted from
320 the assay mixture, approximately 12 or 9%, respectively, residual H₂S-dependent

321 activities were observed compared with the mixture of SoxXA, SoxYZ, SoxB, and
322 SoxCD (34). SoxCD was also required to achieve the maximum rate of
323 H₂S-dependent cytochrome *c* reduction, showing a residual rate of 19% when SoxCD
324 was omitted, compared with the mixture of all Sox enzymes. The results obtained in
325 this study are consistent with those previous results.

326 In *P. pantotrophus* GB17, each *sox* gene encodes a protein with a different
327 molecular mass: SoxX, 16,421 Da; SoxY, 13,830 Da; SoxZ, 11,850 Da; SoxA, 31,884
328 Da; SoxB, 61,897 Da; SoxC, 43,442 Da; and SoxD, 37,637 Da (12). In this study, the
329 estimated molecular masses of rSox enzymes by SDS-PAGE were larger than those
330 predicted by the nucleotide sequences (Fig. 1A). All rSox proteins, except rSoxD, had
331 a 6×His tag, but the differences observed were larger than the size of 6×His
332 (approximately 1.2 kDa). Previous reports (12, 29) also showed that the purified Sox
333 proteins had different molecular masses from predicted ones (SoxX, 16 kDa; SoxY, 12
334 kDa; SoxZ, 16 kDa; SoxA, 29 kDa; SoxB, 59 kDa; SoxC, 47 kDa; and SoxD, 50 kDa)
335 under denaturing SDS-PAGE. Friedrich *et al.* (12) reported that the molecular mass of
336 SoxY determined by SDS-PAGE differed from that predicted by the nucleotide

337 sequence due to an unknown reason. Furthermore, Quentmeier *et al.* (29) also found
338 that the SoxC and SoxD proteins had different molecular masses determined by
339 SDS-PAGE versus those deduced from the nucleotide sequence of the mature protein.
340 Mobility on SDS-PAGE gels may be affected by amino acid composition,
341 glycosylation, or phosphorylation of the protein. In this study, the amino acid
342 composition and predicted pI value were obtained using the Compute pI/Mw tool in
343 ExPASy. None of the rSox enzymes seemed to have a unique amino acid
344 composition, and the pI value was between 4.5 and 6.6. These characteristics do not
345 seem to affect the mobility of the proteins on SDS-PAGE gels. BLAST nucleotide
346 sequence analysis showed the identities of all of the *sox* genes of *P. pantotrophus* GB17
347 used in this study, so the reason for the differences in mass observed remains unclear.

348 The optimal pH for the H₂S-degrading activity of rSox enzymes was 7.0, and the
349 activity was stable at 25-80°C. *P. pantotrophus* was isolated as *Thiosphaera*
350 *pantotropha* originally (33), was reclassified as *Paracoccus denitrificans* (20), and then
351 renamed (31). *T. pantotropha* GB17 obtains energy through the Sox system and can
352 grow at pH 6.5-10.5, with an optimum pH of 8.0, and at a temperature of 15-42°C, with

353 an optimum at 37°C (33). Another report showed optimal growth conditions of *P.*
354 *denitrificans* at pH 7.5-8.0 and 30-37°C (11). These findings suggest that the Sox
355 system functions under these pH and temperature conditions, which seem reasonable.
356 We also found that whole cells from *P. pantotrophus* GB17 could degrade H₂S even
357 after heating at 100°C (data not shown). Recently, thermophilic sulfur-oxidizing
358 bacteria have been isolated from geothermal fields (25, 39). Although it is not clear
359 whether these bacteria contain the Sox system, Sox enzymes might have thermophilic
360 properties. Indeed, Ghosh *et al.* (14) suggested that the Sox system originated in
361 ancient thermophilic bacteria and evolved through extensive horizontal gene transfer.
362 Further studies are necessary to clarify the mechanism of the thermal stability of the
363 rSox enzymes. In possible clinical applications, heat stability of rSox enzymes would
364 be advantageous, because the intra-oral temperature changes after ingestion of various
365 cold/hot meals and beverages (4, 22).

366 In this study, we demonstrated that rSox enzymes significantly degraded H₂S
367 produced by periodontopathic bacteria. The reducing effect of rSox enzymes in
368 reaction mixtures with *P. gingivalis* or *F. nucleatum* was smaller than that obtained in

369 reaction mixtures with NaHS. We examined the effect of rSox enzymes on the growth
370 of these bacterial cells and found no significant effects (data not shown). rSox
371 enzymes also showed potential to decrease the CH₃SH produced by *P. gingivalis*. The
372 Sox system, reconstituted from SoxXA, SoxYZ, SoxB, and SoxCD, mediates
373 thiosulfate-, sulfite-, sulfur-, and H₂S-dependent cytochrome *c* reduction(12, 34), but no
374 reports have demonstrated the CH₃SH-dependent reduction. The availability of
375 CH₃SH as a substrate for Sox system needs to be clarified in further studies. We also
376 demonstrated that H₂S produced by oral bacteria collected from human saliva decreased
377 after incubation with rSox enzymes. These results indicate the potential of rSox
378 enzymes to reduce the H₂S produced in the oral cavity.

379 Many randomized controlled trials have reported reducing effects of mouth
380 rinses containing chlorhexidine, cetylpyridinium chloride, or metal ions on oral
381 malodor, while several studies have reported no significant effects (7, 30, 40).
382 Systematic reviews have reported that available data are insufficient to determine
383 precision (5) and concluded that the strength of the recommendation to use these
384 ingredients for oral malodor reduction is graded to be 'weak' (40). It is unclear

385 whether the Sox enzyme-reducing effect on H₂S is superior to existing components or
386 not. Chlorhexidine and cetylpyridinium chloride act to suppress the bacterial growth
387 and they may be accompanied by several adverse effects on oral tissues. The Sox
388 system seems to be advantageous in this point because the target of it is not oral bacteria
389 or oral tissues but H₂S produced by oral bacteria. Furthermore, many studies have
390 reported that H₂S participates in physiological regulation in human body (28, 46, 50).
391 The Sox system has potential to be an effective agent for controlling body systems as
392 H₂S regulator.

393 For the final goal of clinical application of this enzyme system, more data are
394 necessary. Toxicity of Sox enzyme system should be examined and stability of rSox
395 enzymes against proteases present in saliva is also required. Maximum effect of the
396 rSox enzymes mixture appeared 2 h after reaction started (Fig. 3B). In this study, we
397 examined the rSox enzymes activity using assay mixtures in the tube sealed with a
398 silicon plug. Since many components including type and amount of sulfur source, pH
399 of the oral fluids, salivary flow rate, and oxygen levels in saliva and dental plaque affect
400 VSC formation in mouth (23), it is unclear how long it will take to obtain the maximum

401 effect of the rSox enzymes mixture in the oral cavity. We expect the reducing effect of
402 the rSox enzymes to last for long time as possible. The effect lasted for 4 h in the
403 present assay, but it is also unclear how long the effect will last in the oral environment.
404 Further studies are necessary to clarify these points before clinical application.

405 In conclusion, we found that the combination of seven rSox enzymes from *P.*
406 *pantotrophus* GB17 has H₂S-reducing activity. The results suggest that the rSox
407 system could be useful for the prevention of oral malodor.

408

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413

414 **Conflict of Interest**

415 No conflict of interest declared.

416

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558

Table 1. Oligonucleotide primers used in this study

Fragment	Primer	Sequence (5' to 3') ^a
rSoxX	rSoxX-Forward	AG <u>GGATCC</u> ATGAGCAGCCATCTATGG
	rSoxX-Reverse	CTA <u>AGCTT</u> GTCGAGCCTGTAGAGATC
rSoxY	rSoxY-Forward	AG <u>GGATCC</u> ATGATCCTTTCAAGACGC
	rSoxY-Reverse	GCA <u>AGCTT</u> AATCTCCTGTTACTGGAC
rSoxZ	rSoxZ-Forward	AG <u>GGATCC</u> ATGGCAGATGATGCAAAG
	rSoxZ-Reverse	TGA <u>AGCTT</u> GATGTTGCGGCGCTTAGG
rSoxA	rSoxA-Forward	GAG <u>GATCC</u> ATGCCGCGCTTTACCAAG
	rSoxA-Reverse	GGA <u>AGCTT</u> GAAAGCATTGCCCTTTCGA
rSoxB	rSoxB-Forward	AC <u>GATCC</u> ATGATTACCCGACGTGAG
	rSoxB-Reverse	CAA <u>AGCTT</u> AACGCTCCTTCGTGATTG
rSoxCD	rSoxCD-Forward	GT <u>GATCC</u> ATGAAAGACGAGCTCACC
	rSoxCD-Reverse	AGA <u>AGCTT</u> CTGTCTCATGCGTCACTT

^a Nucleotides underlined in each primer sequence show the position of the restriction endonuclease site incorporated to facilitate cloning.

Table 2. Effect of each rSox enzyme on degradation of H₂S generated from NaHS

Sox enzymes added to the assay							H ₂ S (ng/ml) ^a	Ratio to
X	A	Y	Z	B	CD	BSA ^b		control (%)
-	-	-	-	-	-	-	16.9 ± 3.8	
+	+	+	+	+	+	-	6.4 ± 1.1*	37.9
-	-	+	+	+	+	-	11.9 ± 2.5	70.4
+	+	-	-	+	+	-	11.5 ± 2.1	68.0
+	+	+	+	-	+	-	12.0 ± 4.1	71.0
+	+	+	+	+	-	-	12.0 ± 1.5	71.0
-	-	-	-	-	-	+	17.2 ± 4.7	101.8

To determine the role of each rSox component, an enzyme mixture was prepared as shown in the table. Prepared rSox enzyme mixture (0.125 nmol each) was reacted with 20 nmol of NaHS for 2 h at 37°C.

^a Values are the means ± SDs of three independent experiments.

^b BSA, bovine serum albumin (0.125 nmol) was used as a non-sulfur-oxidizing control.

* $P < 0.05$ compared with the control (without rSox enzymes or BSA), as determined

by ANOVA followed by Dunnett's test.

559

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Table 3. rSox enzyme activity to degrade VSCs produced by periodontopathic bacteria

Bacterial strain	H ₂ S (ng/ml) ^a		CH ₃ SH (ng/ml) ^a	
	rSox enzymes		rSox enzymes	
	(-)	(+)	(-)	(+)
<i>F. nucleatum</i> ATCC10953	26.7 ± 2.2	12.1 ± 1.9*	ND	ND
<i>P. gingivalis</i> W83	19.6 ± 4.5	9.1 ± 3.5*	22.0 ± 2.9	15.9 ± 1.8

^a Values are the means ± SDs of three independent experiments.

* $P < 0.05$ compared with the control (without rSox enzymes), as determined by Student's *t*-test.

ND, not determined.

560

Table 4. rSox enzyme activity to degrade H₂S produced by oral bacteria

Subject	H ₂ S (ng/ml) ^a		Ratio (%) (rSox/Control)
	Control	rSox enzymes	
A	23.8 ± 3.8	13.3 ± 4.7*	55.9
B	22.3 ± 6.3	6.5 ± 0.7*	29.1
C	8.1 ± 1.1	2.7 ± 0.2*	33.3

^a Values are the means ± SDs of three independent experiments.

**P* < 0.05 compared with the control (without rSox enzymes), as

determined by Student's *t*-test.

561

562 **Figure legends**

563

564 **Fig. 1** SDS-PAGE (A) and Western blotting (B) analyses of purified rSox proteins.

565 (A) rSox proteins were suspended in SDS-PAGE reducing buffer (1% SDS, 1%

566 2-mercaptoethanol) and heated at 100°C for 3 min. Samples were subjected to

567 SDS-PAGE (15% polyacrylamide), and the gel was stained with Coomassie brilliant

568 blue R-250. Lanes: 1, molecular mass markers; 2, rSoxX (3 µg); 3, rSoxY (3 µg); 4,

569 rSoxZ (3 µg); 5, rSoxA (3 µg); 6, rSoxB (3 µg); 7, rSoxCD (6 µg). (B) rSox proteins

570 on the gel were transferred electrophoretically to a nitrocellulose membrane, and the

571 membrane was reacted with an antibody against 6×His. Lanes: 1, rSoxX (1.5 µg); 2,

572 rSoxY (1.5 µg); 3, rSoxZ (1.5 µg); 4, rSoxA (1.5 µg); 5, rSoxB (1.5 µg); 6, rSoxCD (3

573 µg).

574

575 **Fig. 2** *P. pantotrophus* GB17 activity to degrade H₂S generated from NaHS.

576 Various amounts of cell suspension of *P. pantotrophus* GB17 were incubated with 20

577 nmol NaHS for 2 h at 37°C. Values are the means ± SDs of three independent

578 experiments. * $P < 0.05$ compared with the control (no *P. pantotrophus* GB17 cells),
579 as determined by ANOVA followed by Dunnett's test.

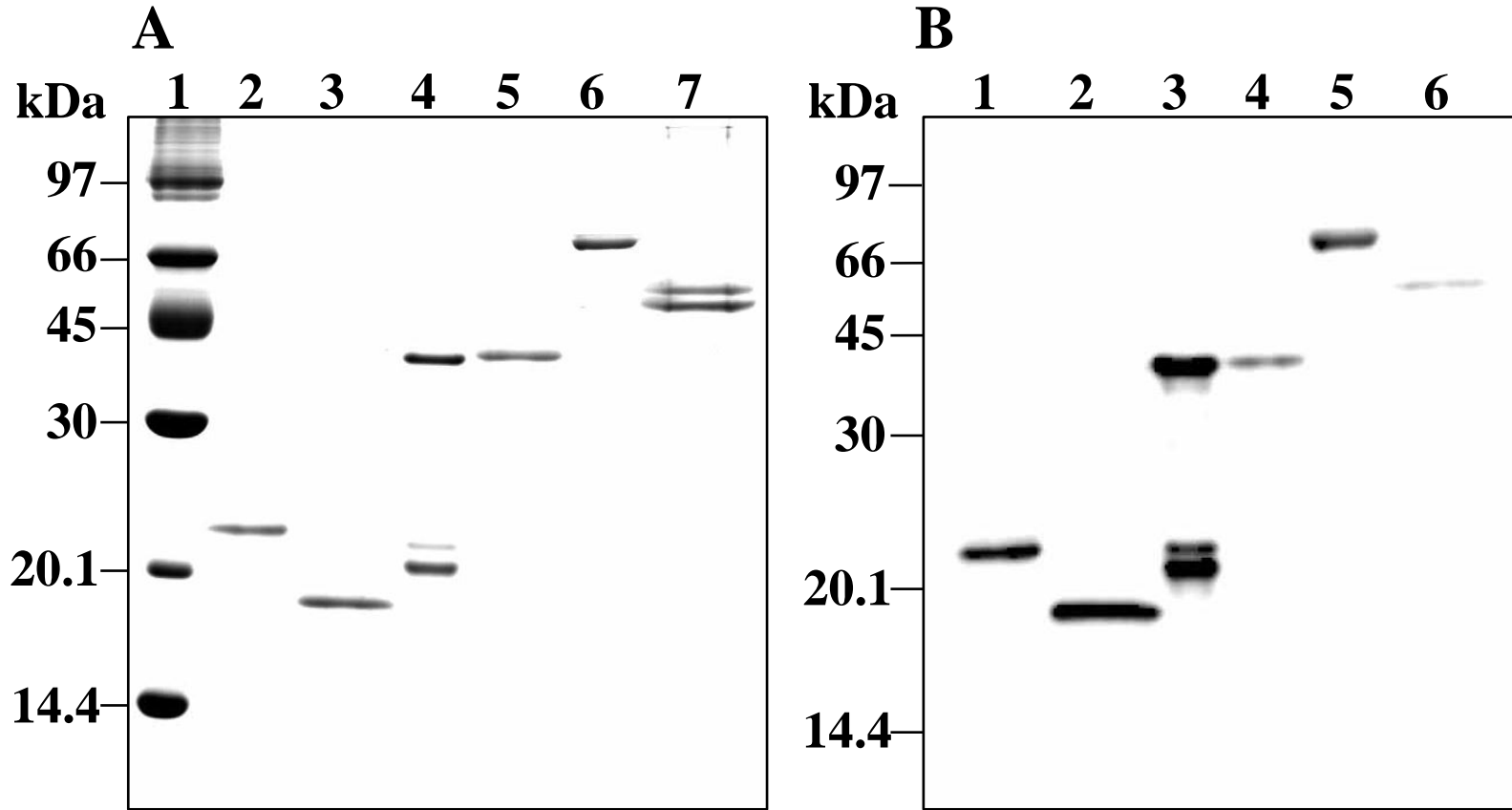
580

581 **Fig. 3** rSox enzyme activity to degrade H₂S generated from NaHS. (A) Various
582 amounts of rSox enzyme mixture were reacted with 20 nmol NaHS for 2 h at 37°C.
583 Values are the means \pm SDs of three independent experiments. * $P < 0.05$ compared
584 with the control (no rSox enzymes), as determined by ANOVA followed by Dunnett's
585 test. (B) rSox enzyme mixture (0.125 nmol each enzyme) was reacted with 20 nmol
586 NaHS for 1 to 4 h at 37°C. The assay was carried out with (sample) or without
587 (control) the rSox enzyme mixture. Values are the means \pm SDs of three independent
588 experiments.

589

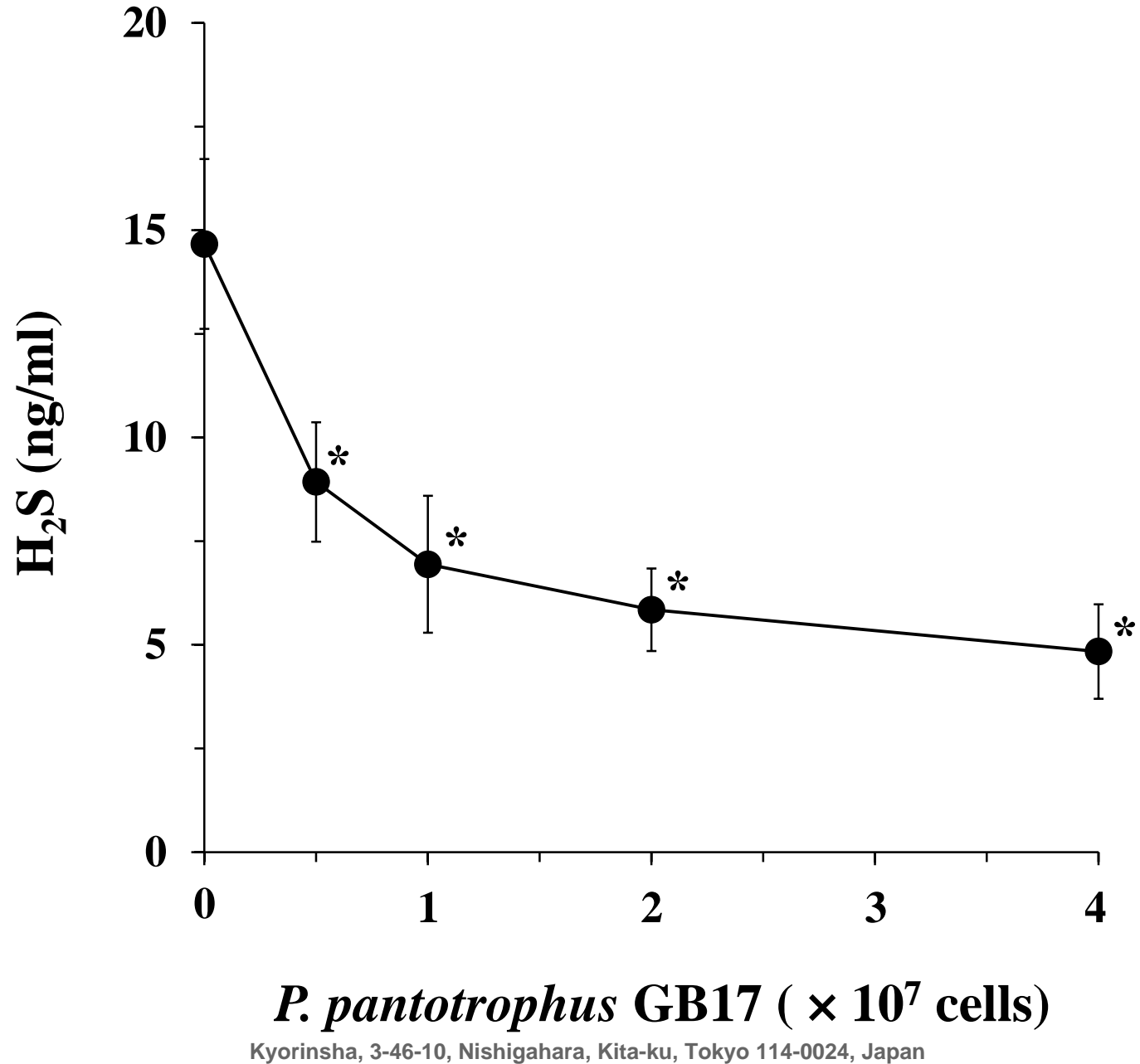
590 **Fig. 4** Heat stability of rSox enzymes (A) and effect of pH on the H₂S-degrading
591 activity of rSox enzymes (B). (A) After rSox enzyme mixture (0.125 nmol each
592 enzyme) was treated at 25 to 100°C for 30 min, the samples were reacted with 20 nmol
593 NaHS for 2 h at 37°C. Open bar, control (no rSox enzymes); solid bars, rSox

594 enzymes. Values are the means \pm SDs of three independent experiments. * $P < 0.05$
595 compared with the control, as determined by ANOVA followed by Dunnett's test. (B)
596 The assay was carried out with (sample) or without (control) rSox enzyme mixture
597 (0.125 nmol each enzyme) in 10 mM phosphate buffer (pH 4 to 9) containing 20 nmol
598 NaHS for 2 h at 37°C. Values are the means \pm SDs of three independent experiments.
599



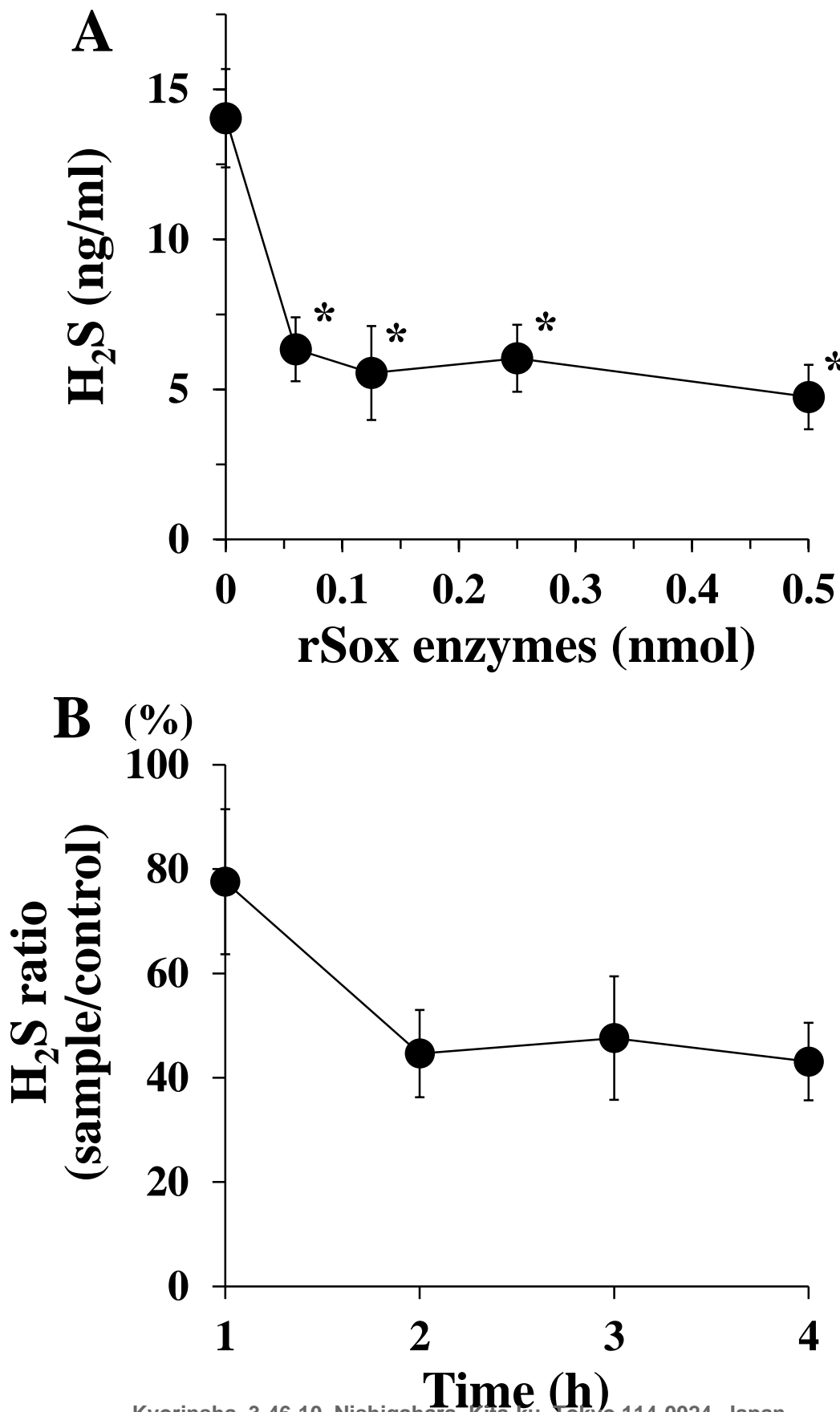
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Fig. 1, Ramadhani



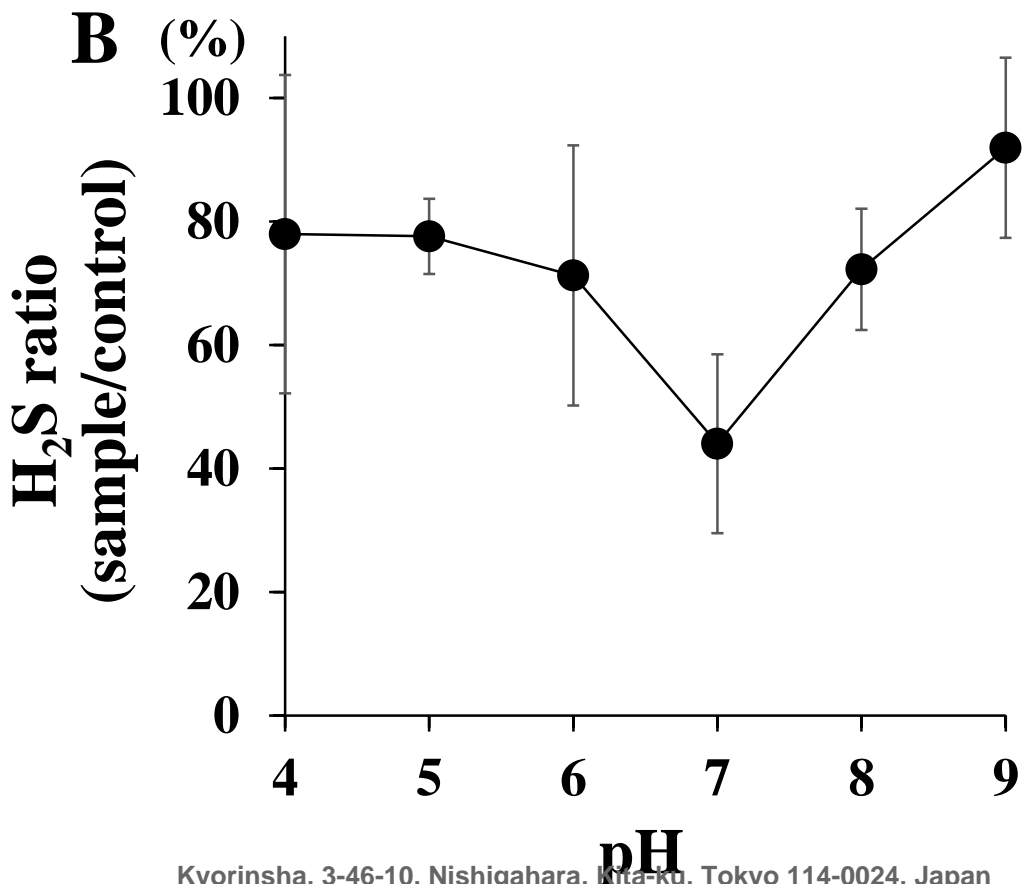
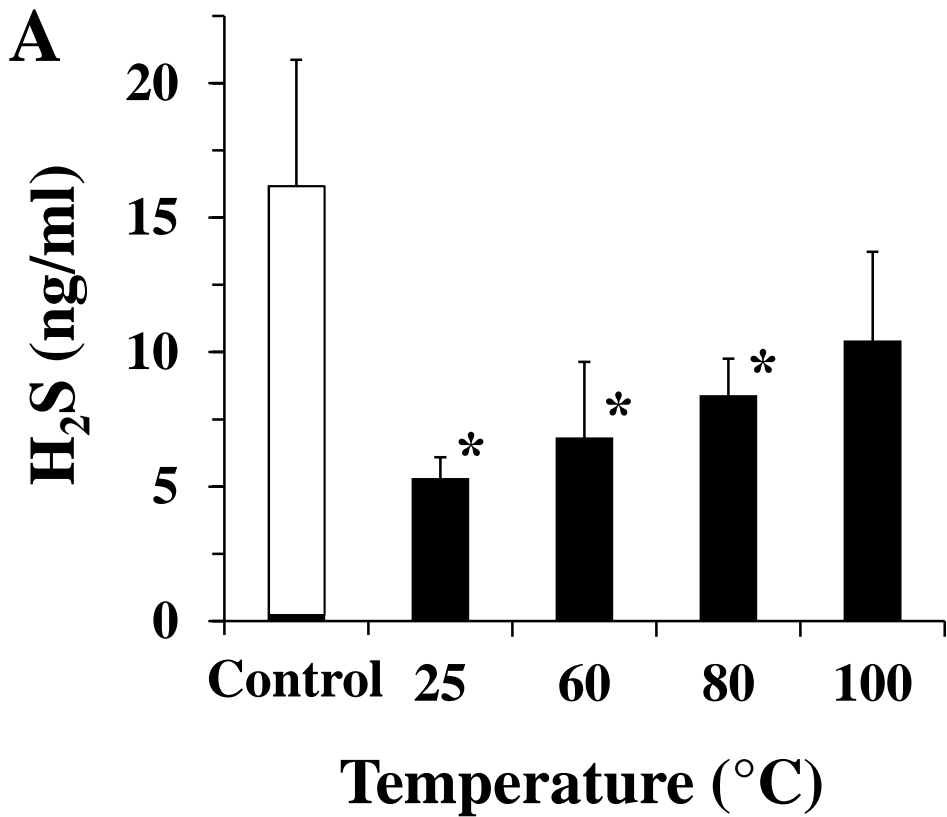
Kyorinsha, 3-46-10, Nishigahara, Kita-ku, Tokyo 114-0024, Japan

Fig. 2, Ramadhani



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Fig. 3, Ramadhani



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Fig. 4, Ramadhani